

Principles of Micro- and Nanofabrication for Electronic and Photonic Devices

Etching 刻蚀 Part III: Dry 干法

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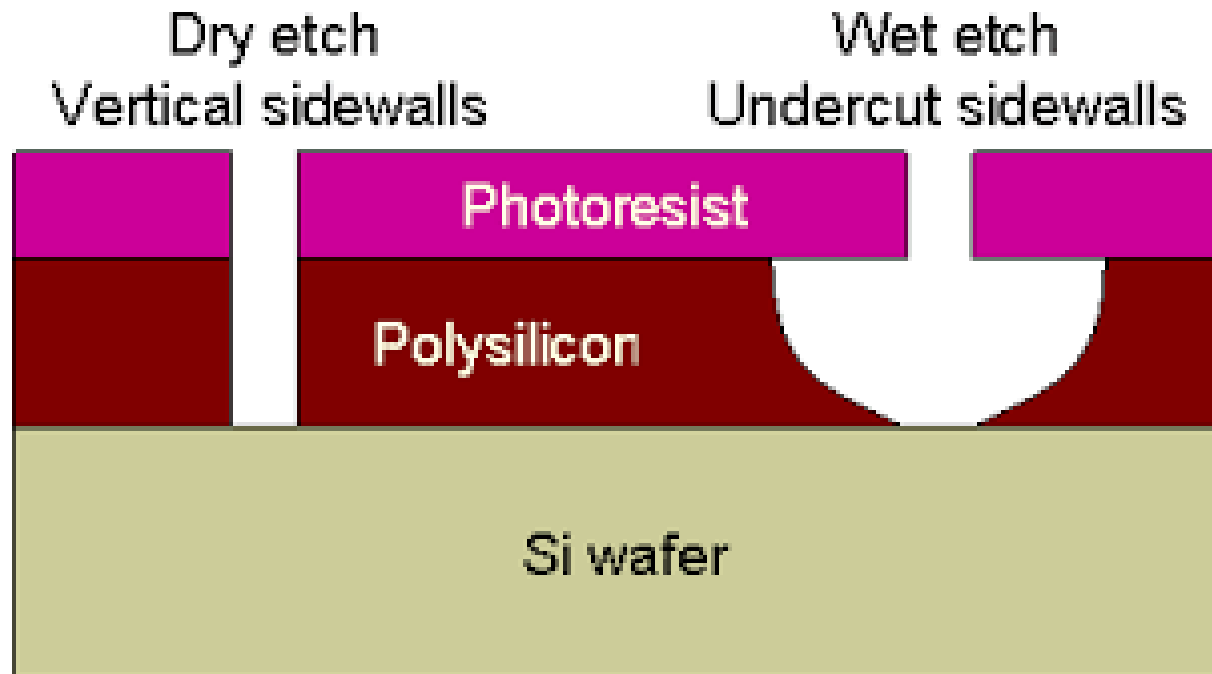
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Etching Methods

- **Wet Etching 湿法刻蚀**
- **Dry Etching 干法刻蚀**
- **CMP and other methods**

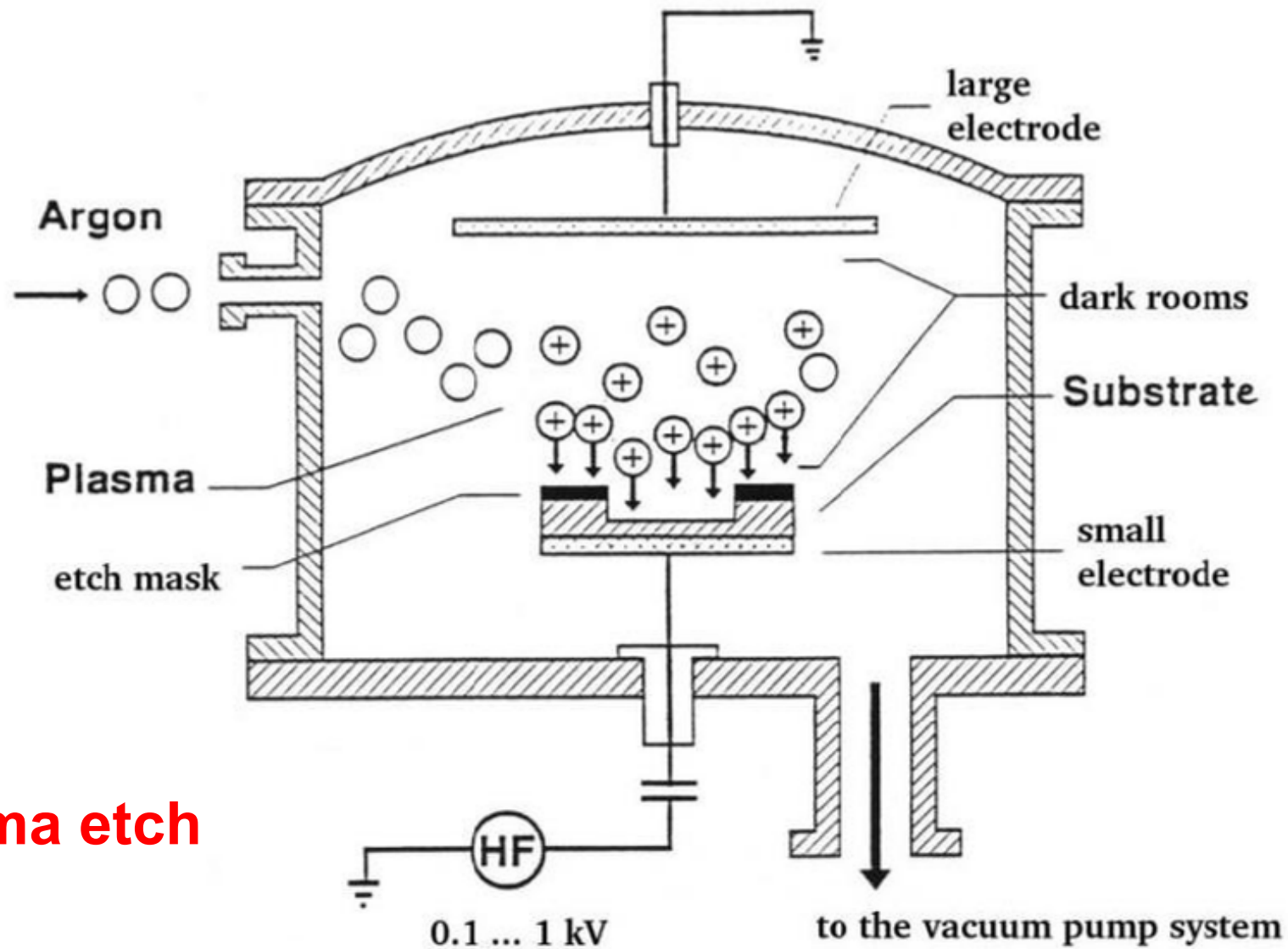
Challenges for Wet Etching

- Most wet etching processes are chemical, isotropic



- For features $< 3 \mu\text{m}$, dry etching has much better resolution

Dry Etching

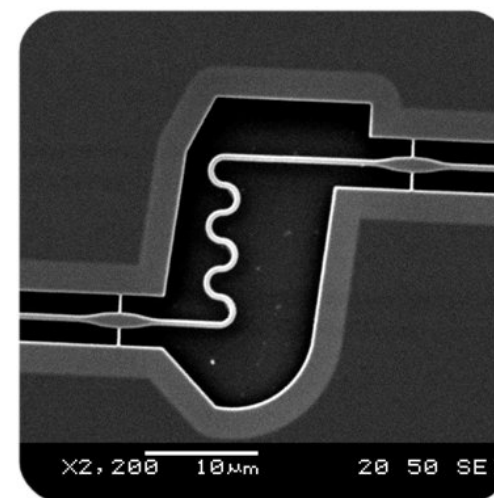
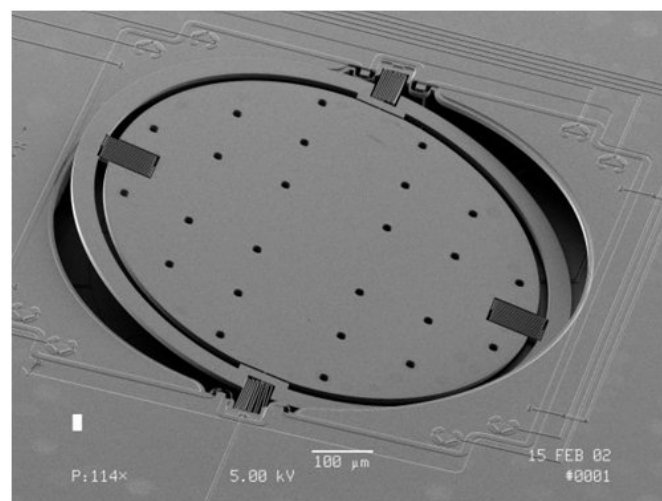
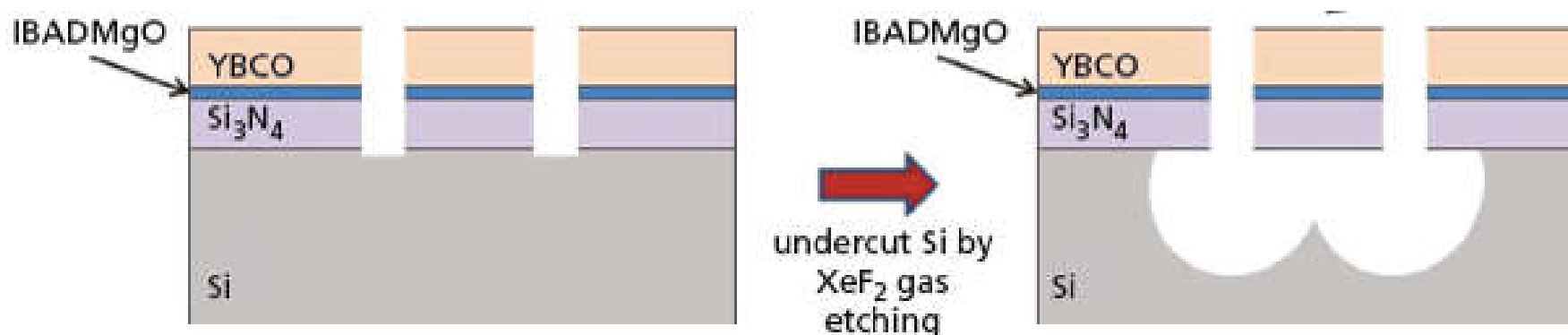


plasma etch

Dry Etch without Plasma

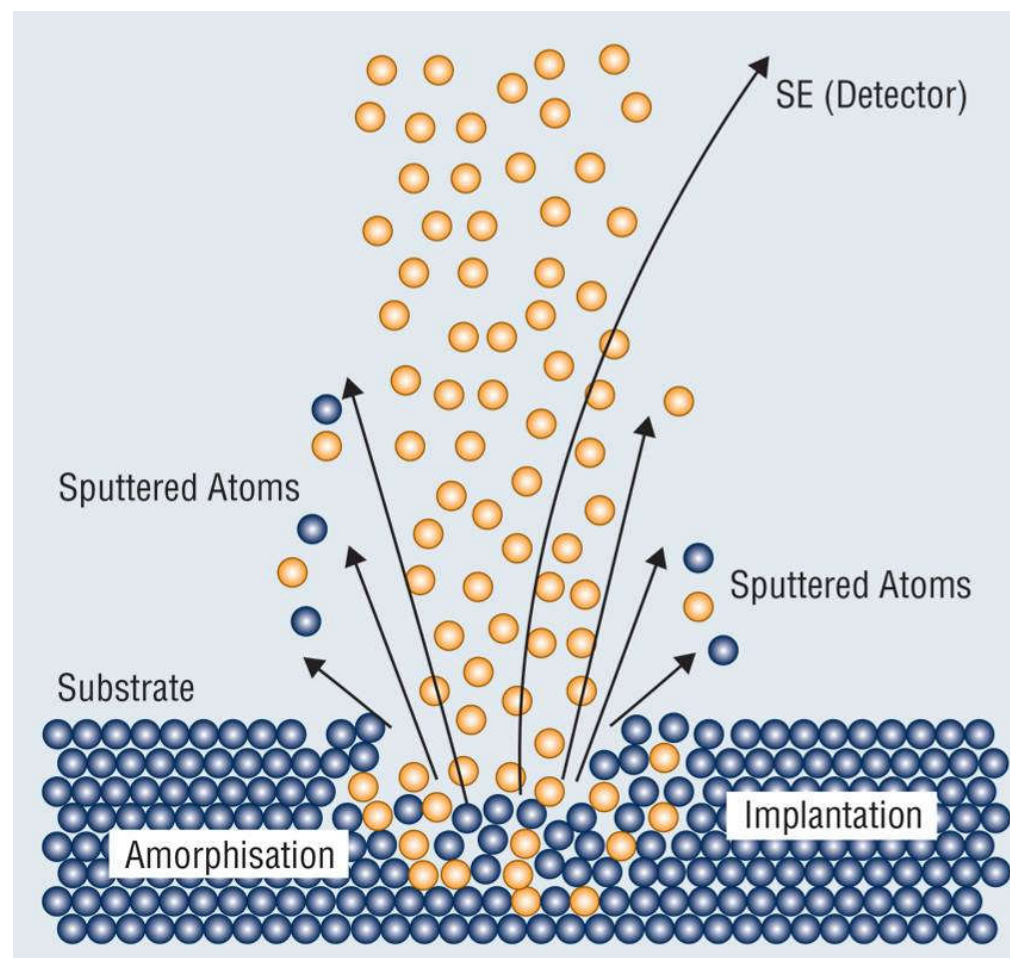
- $2\text{XeF}_2 (\text{g}) + \text{Si} (\text{s}) = 2\text{Xe} (\text{g}) + \text{SiF}_4 (\text{g})$
 - very isotropic and selective

**SiF₄ boiling
point 4 °C**



Ion Milling

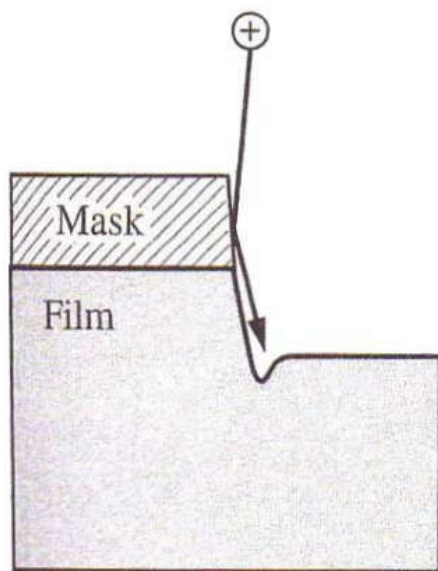
- Heavy ions (e.g. Ar)
- Highly anisotropic
- Poor selectivity
 - for Au, Pt, ...



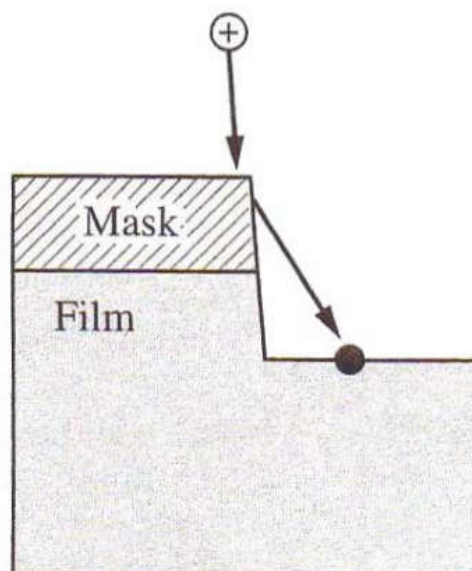
Ion Milling

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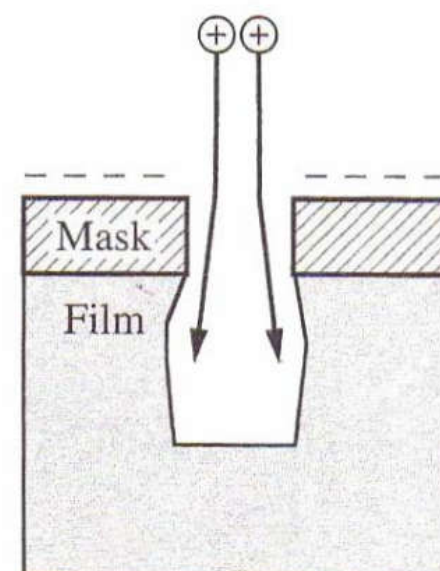
trenching



- mask erosion
- mask redeposition

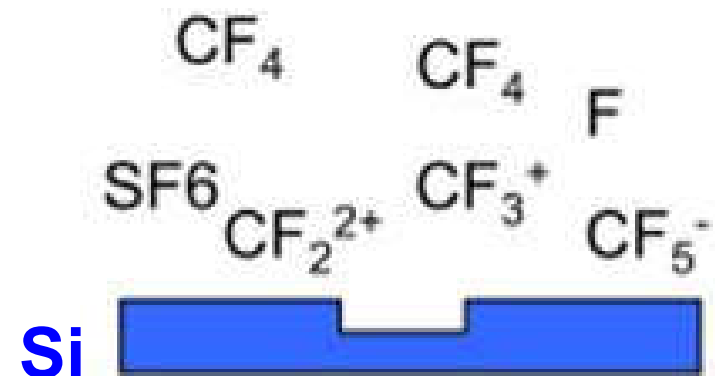


charging of mask:
ion path distortion



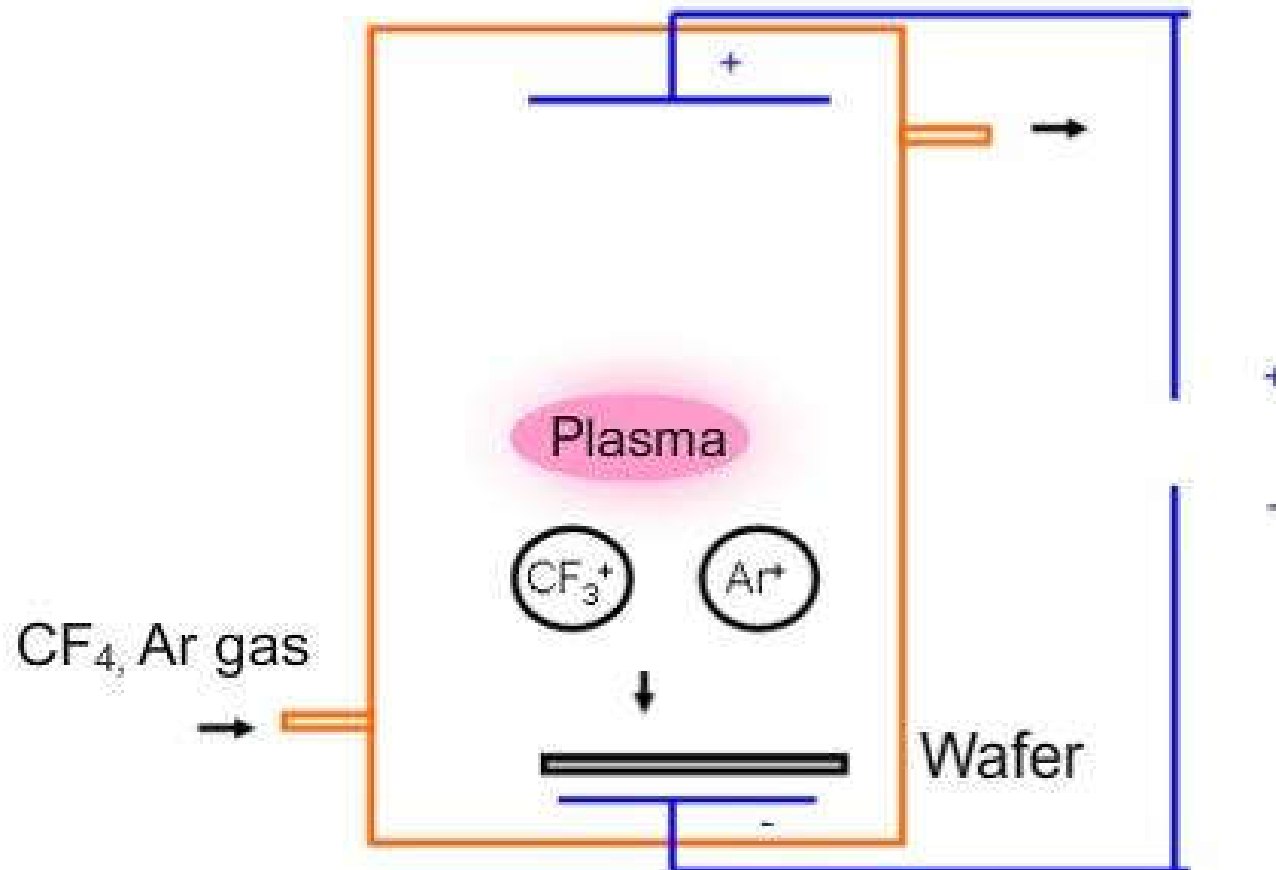
Plasma Etch

- **Chemically reactive ions**
 - improved selectivity
 - lower power
- **Example: Si etch**
 - CF_4 gas does not react with Si
 - energized F^- plasma can react with Si
 - SiF_4 is volatile (boiling point 4 °C)
- **Very isotropic**
 - no direction



Reactive Ion Etching (RIE)

- Improved directionality by applied fields
 - more anisotropic



RIE - Si and SiO₂

- **Si**
 - SF₆ plasma
- **SiO₂**
 - CF₄ / CHF₃ plasma
- **Photoresists can be used as masks**
 - F ions etch PR very slowly

RIE - Si and SiO₂

- Si
 - SF₆ plasma
- SiO₂
 - CF₄ / CHF₃ plasma
- Photoresists can be used as masks
 - F ions etch PR very slowly



SF₆ is heavier than air

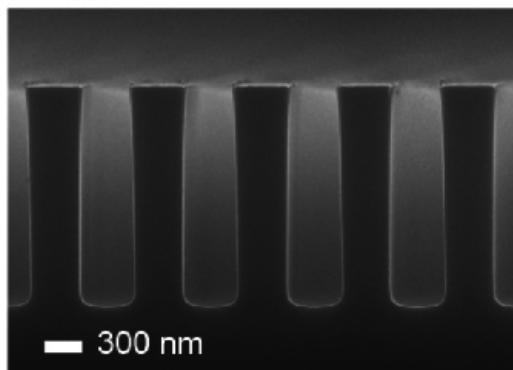
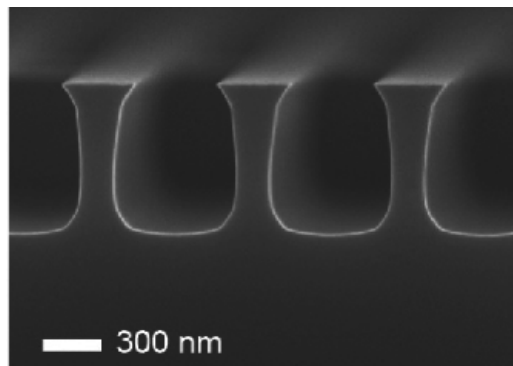
RIE - Organics

- **O₂ plasma**
 - **C-H-O + O⁻ = CO₂ + H₂O**
- **O₂ plasma does not etch Si, SiO₂, or metals**
 - **SiO₂ / metal oxides are non-volatile**

RIE - III-Vs

- Cl_2 / BCl_3 / SiCl_4 plasma
 - GaAs/AlGaAs, InP, GaN/InGaN, ...

Q: why?



GaAs trenches

product	boiling point (°C)
GaF_3	1000
AlF_3	> 1000
InF_3	> 1000
GaCl_3	200
AlCl_3	180
AsCl_3	130

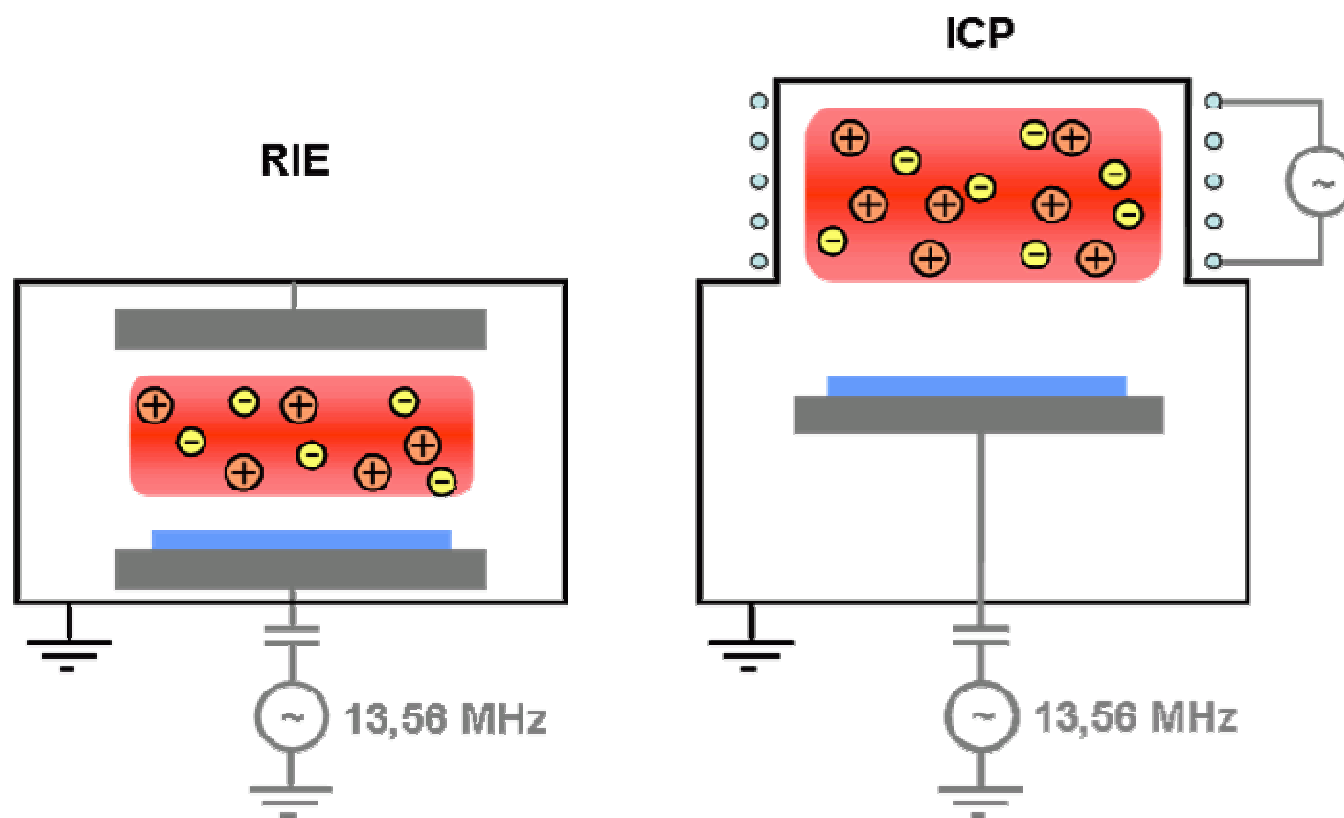
RIE - Recipes

Table 6.2 Materials and corresponding RIE gases

Materials to be etched	Chemical gases (multi choices)
Single-crystal silicon	CF ₃ Br, HBr/NF ₃ , SF ₆ /O ₂
Polysilicon	SiCl ₄ /Cl ₂ , BCl ₃ /Cl ₂ , HBr/Cl ₂ /O ₂ , HBr/O ₂ , Br ₂ /SF ₆
Al	SiCl ₄ /Cl ₂ , BCl ₃ /Cl ₂ , HBr/Cl ₂
Al-Si-Cu, Al-Cu	BCl ₃ /Cl ₂ + N ₂
W	SF ₆ , NF ₃ /Cl ₂
TiW	SF ₆
WSi ₂ , TiSi ₂ , CoSi ₂	CCl ₂ F ₂ /NF ₃ , CF ₄ /Cl ₂
SiO ₂	CCl ₂ F ₂ , CHF ₃ /CF ₄ , CHF ₃ /O ₂ , CH ₃ CHF ₂
Si ₃ N ₄	CF ₄ /O ₂ , CF ₄ /H ₂ , CHF ₃ , CH ₃ CHF ₂
GaAs	SiCl ₄ /SF ₆ , SiCl ₄ /NF ₃ , SiCl ₄ /CF ₄
InP	CH ₄ /H ₂
Photoresists	O ₂

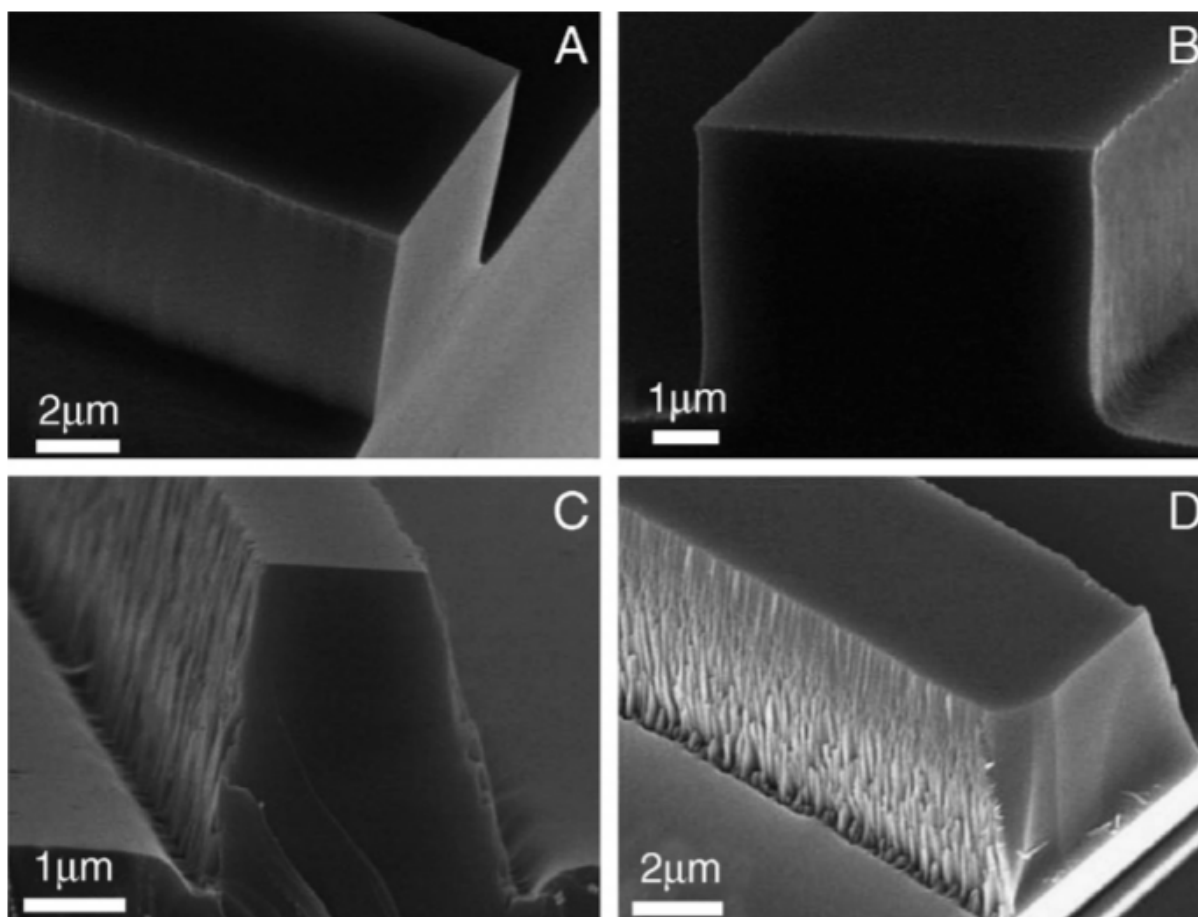
ICP-RIE

- Inductively Coupled Plasma (ICP)
 - higher power
 - mostly for III-Vs



Quality of Etching

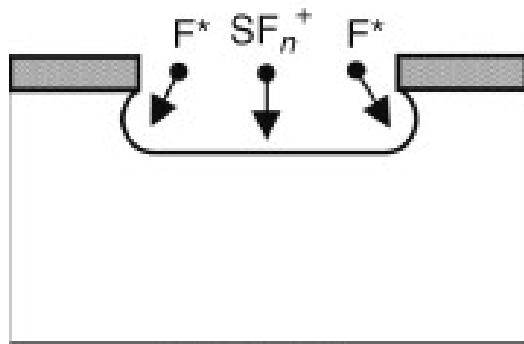
- Quality is controlled by the experiments
 - gas type, pressure, power, ...



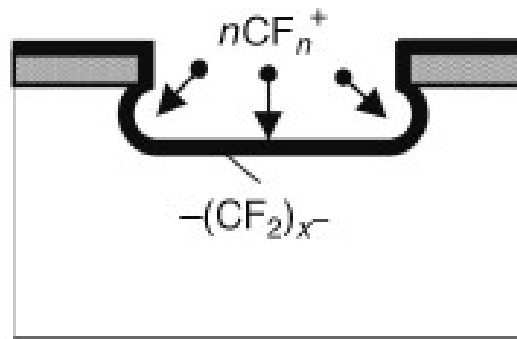
Deep RIE for Si

- alternative etch / passivation

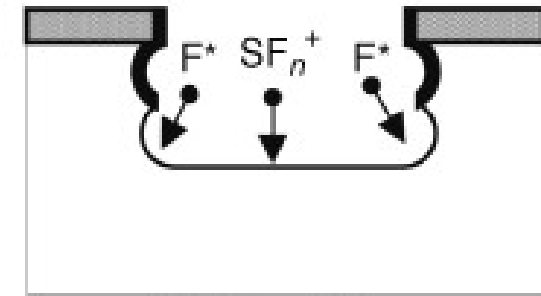
'Bosch process'



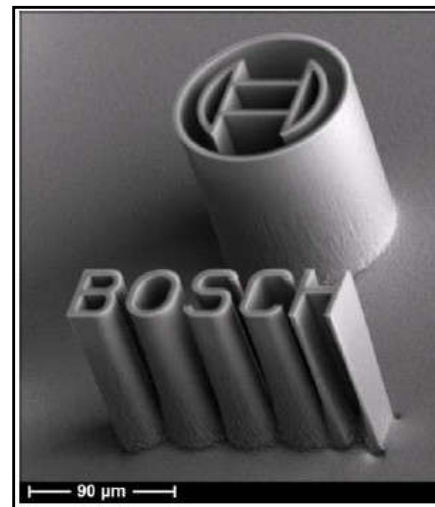
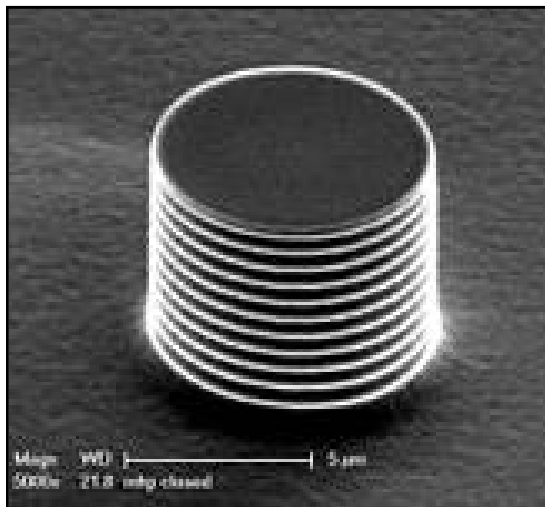
(i) SF₆ etching



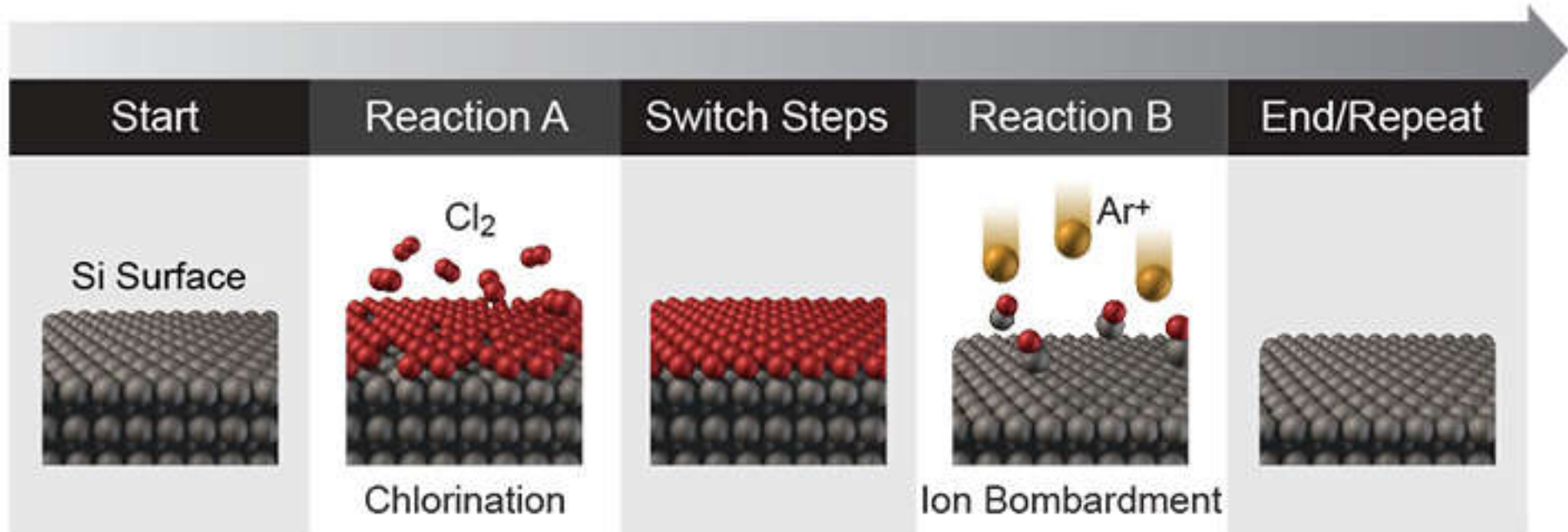
(ii) Passivation



(iii) SF₆ etching

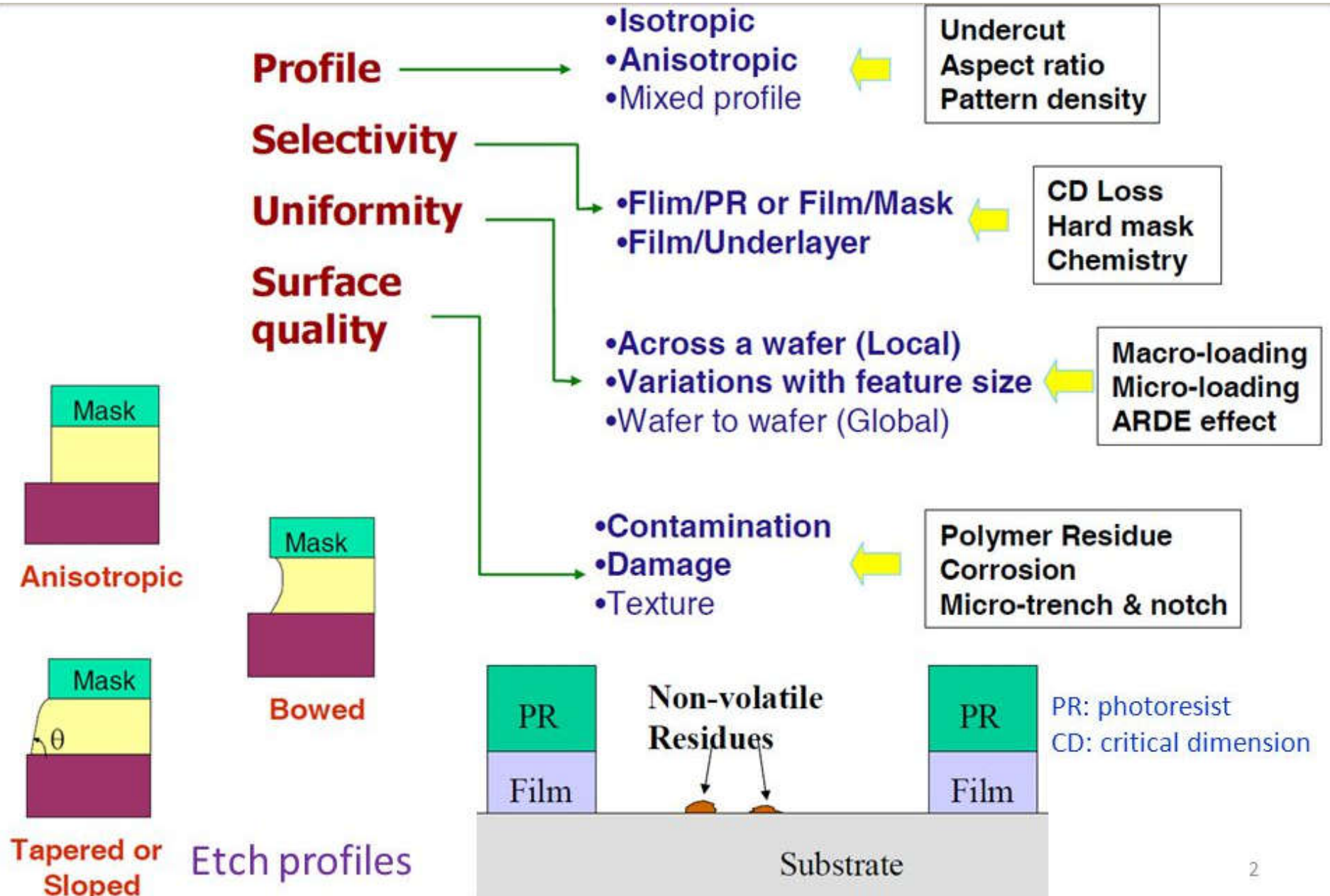


Atomic Layer Etching (ALE)



1. $\text{Si} + 2\text{Cl}_2 = \text{SiCl}_4$
2. SiCl_4 removed by plasma
3. Repeat 1 and 2

Issues in Dry Etch



Summary of Dry Etch

<u>Type of Etching</u>	<u>Excitation Energy</u>	<u>Pressure</u>
Gas/Vapor Etching <i>- isotropic, chemical, very selective</i> (e.g. XeF ₂ gas etch Si even without plasma)	none	high (760-1 torr)
Plasma Etching <i>- isotropic, chemical, selective</i>	10's to 100's of Watts	Medium (>100 torr)
Reactive Ion Etching <i>- directional, physical & chemical, fairly selective</i>	100's of Watts	Low (10-100 mtorr)
Sputter Etching <i>- directional, physical, low selectivity</i> (e.g. ion beam etching/milling using Ar ⁺)	100's to 1000's of Watts	Low (~10 mtorr)